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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/936,531	09/11/2001	Yvan Strauven	U.MINIERE-03	4239
42253	7590 10/14/2004		EXAM	INER
	SHRILAL JAIN 20 MASTERS RUN	CREPEAU, J	ONATHAN	
ELLICOTT CITY, MD 21042			ART UNIT	PAPER NUMBER
			1746	
			DATE MAILED: 10/14/2004	

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)
	09/936,531	STRAUVEN ET AL.
Office Action Summary	Examiner	Art Unit
	Jonathan S. Crepeau	1746
The MAILING DATE of this communication		
Period for Reply	••	
A SHORTENED STATUTORY PERIOD FOR RETHE MAILING DATE OF THIS COMMUNICATIO - Extensions of time may be available under the provisions of 37 CFF after SIX (6) MONTHS from the mailing date of this communication - If the period for reply specified above is less than thirty (30) days, a - If NO period for reply is specified above, the maximum statutory period for reply within the set or extended period for reply will, by strong reply received by the Office later than three months after the mearned patent term adjustment. See 37 CFR 1.704(b).	DN. R 1.136(a). In no event, however, may to reply within the statutory minimum of riod will apply and will expire SIX (6) No atute, cause the application to become	y a reply be timely filed thirty (30) days will be considered timely. MONTHS from the mailing date of this communication.
Status		•
1) Responsive to communication(s) filed on 0:	4 August 2004	
_	This action is non-final.	
3) Since this application is in condition for allo		latters prosecution as to the morite in
closed in accordance with the practice unde		
Disposition of Claims	, , , , , , , , , ,	
·	·	
4) Claim(s) <u>1-17</u> is/are pending in the application		
4a) Of the above claim(s) is/are without 5) Claim(s) is/are allowed.	arawn from consideration.	-
, <u> </u>		
6) Claim(s) <u>1-17</u> is/are rejected.		
7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and	-1/a . 1 . 12	
,,,,,,,,,,	d/or election requirement.	
Application Papers		
9) The specification is objected to by the Exam		
10)☐ The drawing(s) filed on is/are: a)☐ a		
Applicant may not request that any objection to t	he drawing(s) be held in abey	vance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the corr	ection is required if the drawi	ng(s) is objected to. See 37 CFR 1.121(d).
11)☐ The oath or declaration is objected to by the	Examiner. Note the attach	ed Office Action or form PTO-152.
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim for forei	ign priority under 35 U.S.C	. § 119(a)-(d) or (f)
a) ☐ All b) ☐ Some * c) ☐ None of:	• • •	
1. Certified copies of the priority docume	ents have been received.	
2. Certified copies of the priority docume		Application No.
3. Copies of the certified copies of the pi		
application from the International Bure		
* See the attached detailed Office action for a li		ot received.
	1 y	
ttachment(s)		
Notice of References Cited (PTO-892)	4) Interview	Summary (PTO-413)
) I Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/0		p(s)/Mail Date Informal Patent Application (PTO-152)
Paper No(s)/Mail Date	6) Other:	
Patent and Trademark Office OL-326 (Rev. 1-04) Office	Action Summary	Part of Paper No./Mail Date 20041005

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DETAILED ACTION

Response to Amendment

1. This Office action addresses claims 1-10 and newly added claims 11-17. Applicant's declaration under 37 CFR 1.132 is sufficient to overcome the rejection over WO '502. However, claims 1-10 remain rejected for the reasons of record over WO '502 in view of JP '379, and claims 11-17 are newly rejected for these reasons. Accordingly, this action is made final.

Claim Rejections - 35 USC § 103

2. Claims 1-17 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 94/19502 in view of JP 1-52379.

Regarding claim 5, WO '502 is directed to an alkaline battery comprising an anode, cathode, and electrolyte (see claim 23 of the reference). Regarding claim 1, the anode comprises a zinc alloy having a composition which anticipates each of the claimed compositions (a), (b), and (c). For example, the alloy disclosed at page 5, line 30, which contains 70 ppm (0.007 wt%) Al and 250 ppm (0.025 wt%) Bi, anticipates alloy (c) of instant claim 1. The alloy disclosed at page 6, line 12 (0.007 wt% Al, 0.025 wt% In) anticipates alloy (a) of instant claim 1. The alloy disclosed at page 6, line 21 (0.003 wt% Al, 0.025 wt% In, 0.025 wt% Bi) anticipates alloy (b) of instant claims 1 and 2. Regarding claim 6, the powder comprises metal cemented out of the

electrolyte (see claim 24 of the reference). Regarding claim 1, the powder can be made by a centrifugal atomization process (see page 3, line 30).

WO '502 does not expressly teach that the centrifugal atomization process is carried out in an atmosphere with a relatively low (i.e., <4vol%) oxygen content, as recited in claims 1, 3, 4, 7, 8, 9, and 10.

JP '379 is directed to a zinc alloy powder for an alkaline battery (see abstract). The powder is manufactured by atomizing the molten zinc alloy in a low oxygen concentration (<4vol%) atmosphere (see abstract).

Therefore, the invention as a whole would have been obvious to one of ordinary skill in the art at the time the invention was made because the disclosure of JP '379 would motivate the artisan to conduct the centrifugal atomization of WO '502 in an atmosphere containing less than 4 vol% oxygen. In the abstract, JP '379 teaches that the purpose of this atmosphere is "to retard hydrogen gas evolution in spite of a low mercury content." Accordingly, the artisan would be motivated to conduct the centrifugal atomization of WO '502 in an atmosphere containing less than 4 vol% oxygen.

Response to Arguments

3. Applicant's arguments filed August 4, 2004 have been fully considered but they are not persuasive insofar as they apply to the present rejection. Applicant asserts that the skilled person would not be motivated to look to a reference disclosing the production of Hg-amalgamated

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powders (JP '379) to modify a reference teaching non-amalgamated powders (WO '502). However, as noted above, in the abstract, JP '379 teaches that the purpose of the low oxygencontent atmosphere is "to retard hydrogen gas evolution in spite of a low mercury content." Therefore, the abstract fairly suggests that the low oxygen-content atmosphere is the source of the improvement of the powder of JP '379. The artisan would thus be sufficiently motivated to apply this teaching to similar powders in hopes of obtaining a similar improvement. In fact, the abstract of JP '319 appears to teach away from using a large amount of mercury. Thus, the difference in mercury content between the two references is not seen as a critical issue that would dissuade an artisan from combining the references. Similarly, the other elements in the composition of JP '379, e.g., gallium, are also not seen as a critical issue when combining the references. As stated above, it is believed that JP '379 provides sufficient motivation to apply the teaching of low content-atmosphere to any zinc alloy having a similar composition. The Examiner would look favorably upon results showing an unexpected improvement in the claimed compositions made by the claimed method, as opposed to similar compositions (e.g., those of JP '379 or some of those of WO '502) also made by the claimed method. In other words, Applicants should show that their specific composition in addition to the low oxygen-content atmosphere produces an unexpected result. Currently, because the prior art fairly suggests a lowoxygen atmosphere for a variety of zinc alloys, merely relying on results wherein a low-oxygen atmosphere is the only variable is not sufficient to overcome the outstanding rejection.

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As an additional note, a full translation of JP '379 has been ordered and should be available within a few weeks. If Applicants wish to review the translation prior to responding to this Office action, they are respectfully requested to contact the Examiner.

Conclusion

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy 4. as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan Crepeau whose telephone number is (571) 272-1299. The examiner can normally be reached Monday-Friday from 9:30 AM - 6:00 PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Michael Barr, can be reached at (571) 272-1414. The phone number for the organization where this application or proceeding is assigned is (571) 272-1700. Documents may be faxed to the central fax server at (703) 872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Jonathan Crepeau Primary Examiner

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October 8, 2004